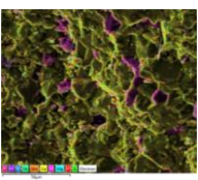
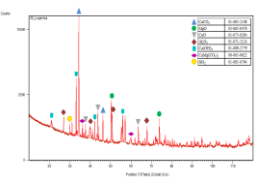
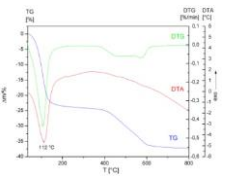
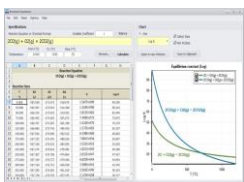
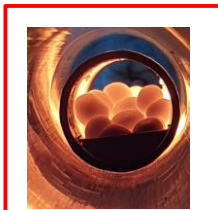




Comprehensive material research of Fe and Mn ores and assessment of possibilities for their reduction with hydrogen.

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Methodologies for material research of Fe and Mn ores

				
chemical composition	physical properties	microstructure of grains	XRD composition	intervals of melting
				possibilities for reduction with hydrogen
DTG, DTA analysis	high – temperature of stability	thermodynamics models	reducibility	

Material research of Fe and Mn ores

Iron ore				
		Krivbas (Ukraine)	Rudomain (Ukraine)	Carajas (Brazil)
Chemical composition (wt%)	Fe TOTAL	62.25	58.20	65.23
	SiO ₂	7.80	13.40	2.21
	P	0.03	0.06	0.05
	S	0.03	0.01	0.01
Mineralogical composition	XRD	hematite, hydrohematite, quartz	hematite, hydrohematite, hydrated iron silicate, cronstedtite, quartz	hematite, quartz
Melting point (°C)		1472	1453	1540

Manganese ore					
		Burkina Faso (Africa)	Gabon (Africa)	Ghana (Africa)	RSA (Africa)
Chemical composition (wt%)	Mn TOTAL	52.05	53.09	37.73	34.32
	SiO ₂	4.66	3.94	5.02	5.61
	P	0.07	0.04	0.08	0.03
Mineralogical composition	XRD	pyrolusite, manganite, pyrochroite, quartz	pyrolusite, quartz, magnetite	rhodochrosite, quartz	rhodochrosite, braunite, quartz, hematite
Melting point (°C)		> 1550	> 1550	1470	1410

The most suitable ores for pilot experiments of hydrogen reduction

iron ore



Carajas

manganese ore



Gabon

Carajas contains easily reducible phase – hematite, high quality, high stability

Gabon contains easily reducible phase – pyrolusite, high quality, high stability

